

## Contents to Volume 1

### List of Contributors XV

1	<b>Introduction</b>	1
	<i>Stefan Kaskel</i>	
2	<b>Network Topology</b>	5
	<i>Frank Hoffmann and Michael Fröba</i>	
2.1	Introduction	5
2.2	Crystal Structures and MOFs Regarded as Nets	6
2.3	Some Introductory Remarks about Graphs, Topology, and Symmetry	7
2.3.1	Interplay of Local Geometry and Topology	10
2.4	Nomenclature of and Symbols for Nets – or What Does 4.4.4.4.4.4.4.4.4.*.*.* Mean?	11
2.4.1	Schläfli Symbols	12
2.4.2	Vertex Symbols	13
2.4.3	Point Symbols	17
2.4.4	Face Symbols	17
2.4.5	RCSR Symbols	18
2.5	Characterization of Nets in the Spirit of the RCSR	21
2.5.1	Pictures, Names, Keywords, and References	21
2.5.2	Embedding and Realization: How to Assign Coordinates to Vertices?	21
2.5.3	Genus	22
2.5.4	Topological Density	24
2.5.5	Specifying the Vertices and Edges of the Net	24
2.5.6	Tilings, Special Form of Pores, and the Concept of Transitivity	25
2.6	Derived Nets: Relationships Between Nets	28
2.7	Simplification of MOFs and Linkers or the Famous Case “4-c versus 2 Times 3-c”	30
2.8	A Very Short Survey on Common and Not So Common Nets: The Minimal Transitivity Principle	34
2.9	Summary and Conclusions	37

Acknowledgment 37

References 38

**Part I MOF Chemistry of Metallic Clusters and Other Nodes 41**

**3 Reticular Chemistry of Metal–Organic Frameworks Composed of Copper and Zinc Metal Oxide Secondary Building Units as Nodes 43**

*Alexander Schoedel and Omar M. Yaghi*

3.1 Introduction 43

3.2 Secondary Building Units (SBUs): The Design Principles of MOFs 43

3.3 Points of Extension 44

3.3.1 Three Points of Extension 45

3.3.2 Four Points of Extension 46

3.3.3 The Discovery and Importance of Open Metal Sites 47

3.3.4 Six Points of Extension 51

3.3.5 Eight Points of Extension 55

3.3.6 Nine Points of Extension 56

3.3.7 Twelve Points of Extension 57

3.3.8 Twelve or Twenty-Four Points of Extension: Metal–Organic Polyhedra 61

3.3.9 Infinite Secondary Building Units 65

3.4 Concluding Remarks 69

Acknowledgment 69

References 69

**4 Alkaline Earth Metal-Based Metal–Organic Frameworks: Synthesis, Properties, and Applications 73**

*Debasis Banerjee, Hao Wang, Benjamin J. Deibert, and Jing Li*

4.1 Introduction 73

4.2 Synthesis 74

4.2.1 Types of Synthesis: Hydrothermal, Solvothermal, and Ionothermal Routes 74

4.2.2 Types of Linkers: Carboxylate, Phosphonate, and N-based Ligands 75

4.3 Structures 76

4.3.1 Beryllium-Based AE-MOFs 76

4.3.2 Magnesium-Based AE-MOFs 77

4.3.3 Calcium-Based AE-MOFs 80

4.3.4 Strontium-Based AE-MOFs 82

4.3.5 Barium-Based AE-MOFs 84

4.3.6 Structural Trends in AE-MOFs 85

4.4 Properties and Applications 86

4.4.1 Gas Storage and Separation 86

4.4.2	Catalysis	93
4.4.3	Luminescence-Based Chemical Sensing	94
4.4.4	Proton and Ionic Conductivity	96
4.5	Conclusions and Outlook	97
	Acknowledgments	98
	List of Abbreviations	98
	References	99
<b>5</b>	<b>Synthesis, Structure, and Selected Properties of Aluminum-, Gallium-, and Indium-Based Metal–Organic Frameworks</b>	<b>105</b>
	<i>Lars-Hendrik Schilling, Helge Reinsch, and Norbert Stock</i>	
5.1	Introduction	105
5.2	Properties of Al <sup>3+</sup> , Ga <sup>3+</sup> , and In <sup>3+</sup> ions	106
5.3	Synthesis and Characterization of G13-MOFs	109
5.4	Prevalent Framework Structures and Inorganic Building Units in G13-MOFs	110
5.5	Selected G13-MOFs	119
5.5.1	Aluminum	119
5.5.2	Gallium	121
5.5.3	Indium	122
5.6	Selected Aspects of G13-MOFs	125
5.6.1	Recent Developments in Synthesis	125
5.6.2	Breathing	127
5.6.3	Postsynthetic Modification	128
	References	130
<b>6</b>	<b>Group 4 Metals as Secondary Building Units: Ti, Zr, and Hf-based MOFs</b>	<b>137</b>
	<i>Mathieu Bosch, Shuai Yuan, and Hong-Cai Zhou</i>	
6.1	Introduction	137
6.1.1	Titanium-Based MOFs	139
6.1.2	MIL-125	141
6.1.2.1	MIL-125 Derivatives and Studies	142
6.2	Zirconium-Based MOFs	144
6.2.1	Introduction	144
6.2.2	UiO-66 and Derivatives	145
6.2.2.1	Stability of the UiO-66 Series	146
6.2.2.2	UiO-66 Titanium Postsynthetic Exchange	147
6.2.2.3	The UiO-66 Series and Modulated Synthesis	148
6.2.2.4	UiO-66 Series Applications	151
6.2.3	Zirconium Porphyrinic MOFs	156
6.2.4	Toward Rationally Designed Zr-MOFs	163
6.2.4.1	PIZOFs	163
6.2.4.2	Zr(IV) Based Metal–Organic Frameworks with Intentionally Altered Topology	163

6.2.4.3	NU-1000	164
6.3	Summary and Conclusions	166
	References	168
<b>7</b>	<b>Iron and Groups V- and VI-based MOFs</b>	<b>171</b>
	<i>Christian Serre and Thomas Devic</i>	
7.1	Introduction: The Chemistry in Solution	171
7.2	MOFs Based on Iron, Chromium, or Vanadium	172
7.2.1	Metal Phosphonates	172
7.2.2	Metal Carboxylates	174
7.2.3	Other Ligands	178
7.3	MOFs Based on Nb, Ru, Mo, and W	179
7.4	Synthesis at the Nanoscale	181
7.5	Properties	182
7.5.1	Flexible MOFs	182
7.5.2	Mechanical Properties	183
7.5.3	Analysis of Their Acidic Behavior	184
7.5.4	Stability Issues	185
7.5.5	Bioapplications of MOFs	186
7.5.6	Redox Properties	188
7.5.7	Catalytic Properties	189
7.5.8	Inclusion	190
7.5.9	Adsorption/Separation	192
7.5.9.1	Gas Storage	192
7.5.9.2	Fluid Separation	192
7.6	Conclusion	194
	Acknowledgments	195
	References	195
<b>8</b>	<b>Platinum Group Metal–Organic Frameworks</b>	<b>203</b>
	<i>Elisa Barea, L. Marleny Rodríguez-Albelo, and Jorge A. R. Navarro</i>	
8.1	Introduction	203
8.2	Single Node Frameworks	204
8.3	Metalloligands for the Construction of Mixed Metal–Organic Frameworks (M'MOFs)	207
8.4	Hofmann-Type MOFs	214
8.5	Coordination Polymers with Paddle-Wheel Metal Clusters	221
8.5.1	One-Dimensional Paddle-Wheel Coordination Polymers	222
8.5.2	Two-Dimensional Paddle-Wheel Coordination Polymers	224
8.5.3	Three-Dimensional Paddle-Wheel Coordination Polymers	225
8.6	Summary and Conclusions	227
	References	227

9	<b>Group 3 Elements and Lanthanide Metals</b>	231
	<i>Klaus Müller-Buschbaum</i>	
9.1	Introduction	231
9.2	Chemistry and Structures of Group 3 and Lanthanide-Based MOFs	232
9.2.1	Relevant Chemical Properties and Background of Group 3 and Lanthanide Elements	232
9.2.2	Synthesis and Design of Group 3 and Ln-MOFs	235
9.2.3	Carboxylate and Noncarboxylate Group 3 and Ln-MOFs	236
9.3	Electronic and Optical Properties of Group 3 and Lanthanide-Based MOFs	246
9.3.1	Electronic Properties of Group 3 and Lanthanide Ions	246
9.3.2	Luminescence of Group 3 and Ln-MOFs	249
9.3.3	Sensing by Luminescence of Group 3 and Ln-MOFs	257
9.4	Summary and Conclusions	263
	References	264

## Part II Functional Linkers 271

10	<b>Extended Linkers for Ultrahigh Surface Area Metal–Organic Frameworks</b>	273
	<i>Hiroyasu Furukawa and Xixi Sun</i>	
10.1	Introduction	273
10.2	Brief Introduction of the History of Porous MOFs	273
10.2.1	Design of Ultrahigh Porosity	273
10.2.2	Use of Highly Porous MOFs	279
10.3	General Synthetic Strategy for Extended Organic Linkers	280
10.3.1	Designing Linker Synthesis	280
10.3.2	Core Unit	283
10.3.3	Extending Units	283
10.3.4	Construction of Linker Backbones	283
10.3.5	Introduction of Binding Sites into Linkers	285
10.4	Case Studies of Extended Linkers	285
10.4.1	Ditopic Linkers	285
10.4.2	Tritopic Linkers	289
10.4.3	Tetratopic Linkers	293
10.4.4	A Pentatopic Linker	298
10.4.5	Hexatopic Linkers	298
10.4.6	Octatopic Linkers	301
10.4.7	Dodecatopic Linkers	302
10.5	Summary and Conclusions	303
	Acknowledgment	304
	References	304

<b>11</b>	<b>Porous Metal Azolate Frameworks</b>	<b>309</b>
	<i>Pei-Qin Liao, Chun-Ting He, Dong-Dong Zhou, Jie-Peng Zhang, and Xiao-Ming Chen</i>	
11.1	Introduction	309
11.2	Imidazolate Coordination Modes	311
11.2.1	Zeolitic and Zeolite-Like Frameworks	311
11.2.2	Polyimidazolates	315
11.3	Pyrazolate Coordination Modes	317
11.3.1	Cluster-Based Networks	318
11.3.2	Chain-Based Pillared-Column Frameworks	324
11.4	Triazolate Coordination Modes	329
11.4.1	1,2,4-Triazolate Coordination Modes	329
11.4.2	1,2,3-Triazolate Coordination Modes	332
11.5	Tetrazolate and Other Coordination Modes	335
11.6	Summary and Conclusions	338
	Acknowledgments	338
	References	338
<b>12</b>	<b>Functional Linkers for Catalysis</b>	<b>345</b>
	<i>Alexandre Legrand, Jérôme Canivet, and David Farrusseng</i>	
12.1	Introduction: MOF in Catalysis	345
12.2	Self-Assembled Frameworks	347
12.2.1	Organocatalyst Ligands	347
12.2.2	Metal-Functionalized Ligands	352
12.3	Postsynthetic Modification	361
12.3.1	Organocatalyst Grafting	361
12.3.2	Postsynthetic Metalation	366
12.4	Relevant and Accurate Characterizations as Key for the Design of MOF Catalyst	376
	List of Abbreviations	381
	References	383
<b>13</b>	<b>Chiral Linker Systems</b>	<b>387</b>
	<i>Christel Kutzscher, Philipp Müller, Silvia Raschke, and Stefan Kaskel</i>	
13.1	Introduction	387
13.2	Section A: Classes of Chiral Linkers	388
13.2.1	Amino Acids and Related Linkers	388
13.2.1.1	Amino Acid Derivatives	389
13.2.1.2	Peptide-Related Linkers	392
13.2.2	Linkers from Natural Sources	396
13.2.3	Rigid Linkers with Chiral Substituents	397
13.2.4	Salen Linkers	399
13.2.5	Linkers with Axial Chirality	401
13.3	Section B: Enantioselective Separation and Chromatography with Chiral MOFs	408

13.3.1	Conceptual Remarks	408
13.3.2	Analytical Separations	410
13.3.2.1	Liquid Chromatography: LC, HPLC, CEC	410
13.3.2.2	GC	413
13.3.3	Preparative Separations	413
13.3.4	Simulation	414
13.4	Summary and Conclusions	415
	References	415
<b>14</b>	<b>Functional Linkers for Electron-Conducting MOFs</b>	<b>421</b>
	<i>Gang Xu, Guo Cong Guo, Ming Shui Yao, Zhi Hua Fu, and Guan E. Wang</i>	
14.1	Introduction	421
14.2	Methods for Measuring Electrical Properties	422
14.3	Linkers	424
14.3.1	Carboxylates	424
14.3.2	Dtao and Its Derivatives	431
14.3.3	Halogens	436
14.3.3.1	MX	436
14.3.3.2	MMX	438
14.3.4	Nitrogen-Containing Heterocyclic Compounds	440
14.3.5	Catechol and Its Derivatives	442
14.3.6	TCNQ and Its Derivatives	446
14.3.7	Cyanide as Linkers	452
14.4	Conclusion and Perspective	456
	Acknowledgments	458
	References	459
<b>15</b>	<b>Linkers with Optical Functionality</b>	<b>463</b>
	<i>Mark D. Allendorf, Kirsty Leong, and Ryan A. Zarkesh</i>	
15.1	Linker Electronic Structure	463
15.2	Design Principles	468
15.2.1	Primary Structure	473
15.2.2	Secondary Structure	473
15.2.3	Tertiary Structure	473
15.2.4	Quaternary Structure	474
15.3	Linkers for Light Harvesting	474
15.3.1	Porphyrin Linkers	476
15.3.2	Organometallic Complexes as Linkers	477
15.3.3	Guest-Facilitated Energy Transfer	478
15.4	Bioimaging	479
15.4.1	Magnetic Resonance Imaging (MRI)	480
15.4.2	Optical Imaging	481
15.4.3	X-Ray Computed Tomography (CT)	482
15.5	Linkers for Chemical Sensing	483

15.6	Radiation Detection	485
15.7	Conclusions	487
	References	487

## Contents to Volume 2

### List of Contributors XIII

#### Part III Special MOF Classes and Morphology Design of MOFs 491

16	Nanoparticles	493
	<i>Michael Beetz, Andreas Zimpel, and Stefan Wuttke</i>	
17	SURMOFs: Liquid-Phase Epitaxy of Metal–Organic Frameworks on Surfaces	523
	<i>Lars Heinke, Hartmut Gliemann, Pierre Tremouilhac, and Christof Wöll</i>	

18	Granulation and Shaping of Metal–Organic Frameworks	551
	<i>U-Hwang Lee, Anil H. Valekar, Young Kyu Hwang, and Jong-San Chang</i>	

#### Part IV Progress in Advanced Characterization of MOFs 573

19	Adsorption Methodology	575
	<i>Irena Senkovska, Katie A. Cychosz, Philip Llewellyn, Matthias Thommes, and Stefan Kaskel</i>	

20	Nuclear Magnetic Resonance of Metal–Organic Frameworks (MOFs)	607
	<i>Stephan I. Brückner, Julia Pallmann, and Eike Brunner</i>	

21	Electron Paramagnetic Resonance	629
	<i>Matthias Mendt, Mantas Šimėnas, and Andreas Pöppl</i>	

22	IR and Raman Spectroscopies Probing MOFs Structure, Defectivity, and Reactivity	657
	<i>Francesca Bonino, Carlo Lamberti, and Silvia Bordiga</i>	

23	In Situ X-ray Diffraction and XAS Methods	691
	<i>Irena Senkovska and Volodymyr Bon</i>	

24	In Situ Studies of the Crystallization of Metal–Organic Frameworks	729
	<i>Richard I. Walton and Franck Millange</i>	

25	<b>Role of Molecular Simulations in the Field of MOFs</b>	765
	<i>Guillaume Maurin</i>	
26	<b>Defects and Disorder in MOFs</b>	795
	<i>Olesia Halbherr and Roland A. Fischer</i>	
A	<b>Appendix A: MOF Suppliers</b>	823
B	<b>Appendix B: Datasheets</b>	825
	<b>Index</b>	833